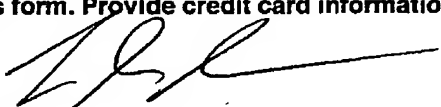


JAN 05 2006

TRANSMITTAL LETTER (General - Patent Pending)				Docket No. YOR920030336US1 (16898)	
In Re Application Of: Timothy J. Dalton, et al.					
Application No. 10/685,012	Filing Date October 14, 2003	Examiner Jose R. Diaz	Customer No. 23389	Group Art Unit 2815	Confirmation No. 4288
Title: PHOTORESIST ASH PROCESS WITH REDUCED INTER-LEVEL DIELECTRIC (ILD) DAMAGE					
<u>COMMISSIONER FOR PATENTS:</u>					
Transmitted herewith is: Declaration Pursuant to 37 C.F.R. § 1.131					
in the above identified application.					
<input checked="" type="checkbox"/> No additional fee is required.					
<input type="checkbox"/> A check in the amount of _____ is attached.					
<input checked="" type="checkbox"/> The Director is hereby authorized to charge and credit Deposit Account No. 50-0510/IBM as described below.					
<input type="checkbox"/> Charge the amount of _____					
<input checked="" type="checkbox"/> Credit any overpayment.					
<input checked="" type="checkbox"/> Charge any additional fee required.					
<input type="checkbox"/> Payment by credit card. Form PTO-2038 is attached.					
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 _____ Signature			Dated: January 5, 2006		
Leslie S. Szivos, Ph.D. Registration No. 39,394 SCULLY, SCOTT, MURPHY & PRESSER 400 Garden City Plaza - Suite 300 Garden City, New York 11530					
<div style="border: 1px solid black; padding: 5px; width: fit-content;"><p>I hereby certify that this correspondence is being deposited with the United States Postal Service with sufficient postage as first class mail in an envelope addressed to the "Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22316-1450" [37 CFR 1.8(a)] on _____</p><p style="text-align: center;">(Date)</p><p style="text-align: center;">_____ Signature of Person Mailing Correspondence</p><p style="text-align: center;">_____ Typed or Printed Name of Person Mailing Correspondence</p></div>					
CC:					

PTO/REV03

JAN 05 2006

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(General - Patent Pending)Docket No.
YOR920030336US1 (16898)

In Re Application Of: Timothy J. Dalton, et al.

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10/685,012	October 14, 2003	Jose R. Diaz	23389	2815	4288

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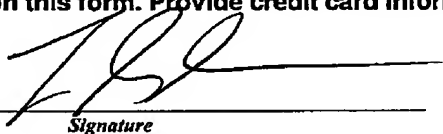
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Dated: January 5, 2006

Leslie S. Szivos, Ph.D.
Registration No. 39,394
SCULLY, SCOTT, MURPHY & PRESSER
400 Garden City Plaza - Suite 300
Garden City, New York 11530

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JAN 05 2006

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE**Applicant(s):** Timothy J. Dalton**Examiner:** Jose R. Diaz**Serial No:** 10/685,012**Art Unit:** 2815**Filed:** October 14, 2003**Docket:** YOR920030336US1 (16898)**For:** PHOTORESIST ASH PROCESS WITH
REDUCED INTER-LEVEL DIELECTRIC (ILD)
DAMAGE**Dated:** January 3, 2006**Confirmation No:** 4288Mail Stop AF
Commissioner for Patents
P.O. Box 1450
Alexandria, VA 22313-1450**DECLARATION PURSUANT TO 37 C.F.R. §1.131**

Sir:

We, Timothy J. Dalton, Nicholas C.M. Fuller, and Kaushik A. Kumar, hereby declare that:

1. We are co-inventors of the subject matter described and claimed in the above-identified patent application.
2. Prior to October 10, 2003, which is the effective date of U.S. Patent Application Publication No. 2005/0077597 A1 to Toma, et al. (hereinafter "Toma"), we have conceived and reduced to practice an interconnect structure that comprises a semiconductor substrate comprising one or more device regions and one or more interconnect levels located atop the semiconductor substrate, wherein the one or more interconnect levels comprises a

patterned organosilicate dielectric having sidewalls that comprise CH_x species with $x = 1-3$, as recited by Claim 1 of the present application.

3. As evidence of the conception and reduction to practice of the claimed interconnect structure referred to in paragraph 2 prior to the effective date of the Toma reference, annexed hereto are Exhibits A and B. Exhibit A is a true photocopy of a write-up for IBM Invention Disclosure YOR8-2003-0269 (hereinafter "Disclosure"), which was created prior to October 10, 2003. Exhibit A describes in general the fabrication of an interconnect structure by using an *in situ* Ar/ H_2 plasma ash process (see Exhibit A, page 1, lines 8-12, and page 3, lines 2-4). Specifically, Exhibit A describes that the interconnect structure comprises patterned organosilicate (OSG) dielectric layer with via and/or trench structures therein (see Exhibit A, page 2, lines 10-18), and that the hydrogen atoms provided during the *in situ* Ar/ H_2 plasma ash process interacts with the OSG sidewall with a reduced chemical reactivity and forms less volatile CH_x ($x = 1-3$) species (see Exhibit A, page 3, lines 10-20). Therefore, Exhibit A establishes clear evidence of actual reduction to practice of the claimed interconnect structure as recited by Claim 1 of the present application. Exhibit B is a true photocopy of an email from Nicholas C.M. Fuller attaching an electronic copy of the Disclosure. The email was sent prior to October 10, 2003, thereby evidencing the actual reduction to practice of the claimed interconnect structure prior to the effective date of the Toma reference. All dates on Exhibits A and B have been blacked out in preparation of this Declaration, but which dates are prior to the October 10, 2003 effective date of the Toma reference.

4. We do hereby declare that all statements made herein of our own knowledge are true, and that all statements made on information and belief are believed to be true; and

further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under 18 U.S.C. § 1001, and that such willful false statements may jeopardize the validity or enforceability of the patent.

Dated

01/04/06
Dated

Dated

Timothy J. Dalton




Nicholas C.M. Fuller

Kaushik A. Kumar

further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under 18 U.S.C. § 1001, and that such willful false statements may jeopardize the validity or enforceability of the patent.

29 Dec Dec 05
Dated


Timothy J. Dalton

Dated

Nicolas C.M. Fuller

Dated

Kaushik A. Kumar

further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under 18 U.S.C. § 1001, and that such willful false statements may jeopardize the validity or enforceability of the patent.

Dated

Timothy J. Dalton

Dated

Nicolas C.M. Fuller

Jan 04, 2006

Dated

Kaushik

Kaushik A. Kumar

EXHIBIT A

U:\Mimi\NBM\105\16898\16898.131declaration.doc

YOR8-2003-0269

PHOTORESIST ASH PROCESS WITH REDUCED INTER-LEVEL DIELECTRIC
(ILD) DAMAGE

5

BACKGROUND

1. Technical Field

The present invention generally relates to integrated circuits (ICs), and more particularly to interconnect structures, including multilevel interconnect structures, in which the original physical and chemical integrity of the dielectric is significantly retained by employing an ash process for photoresist removal post single and dual damascene processing that induces minimal ILD sidewall physical and chemical modification. The present invention is also significant for wafer de-fluorination post barrier (cap) removal during dual damascene processing.

15

2. Description of Related Art

Generally, semiconductor devices include a plurality of circuits which form an integrated circuit including chips (e.g., chip back end of line, or "BEOL"), thin film packages and printed circuit boards. Integrated circuits can be useful for computers and electronic equipment and can contain millions of transistors and other circuit elements that are fabricated on a single silicon crystal substrate. For the device to be functional, a complex network of signal paths will normally be routed to connect the circuit elements distributed on the surface of the device. Efficient routing of these signals across the device can become more difficult as the complexity and number of the integrated circuits are increased. Thus, the formation of multi-level or multi-layered interconnection schemes such as, for example, dual damascene wiring structures, have become more desirable due

25

to their efficacy in providing high speed signal routing patterns between large numbers of transistors on a complex semiconductor chip. Within the interconnection structure, metal vias run perpendicular to the silicon substrate and metal lines run parallel to the silicon substrate.

5 Presently, interconnect structures formed on an integrated circuit chip consists of at least about 2 to 8 wiring levels fabricated at a minimum lithographic feature size designated about 1x (referred to as "thinwires") and above these levels are about 2 to 4 wiring levels fabricated at a width equal to about 2x and/or about 4x the minimum width of the thinwires (referred to as "fatwires"). In one class of structures, the thinwires are
10 formed in a low dielectric constant (k) organosilicate (OSG) dielectric layer, and the fatwires are made in a silicon dioxide dielectric layer having a dielectric constant of about 4.

 However, unlike silicon dioxide ILD structures, there are issues associated with retaining the original chemical and physical integrity of OSG materials as the
15 "thinwire" structures are formed during single and dual damascene processing. Specifically, once the "thinwire" structure is formed, in typical "via-first" integration strategies, it is necessary to subsequently remove on the order of 100 to 300nm of photoresist or organic material. Since the via and/or trench structure is present, the resist ash chemistry employed can potentially interact with the exposed OGS sidewall and
20 modify the material properties (which can also occur in some "trench-first" integration strategies where the OSG dielectric is also exposed to the ash chemistry employed). This modified layer can typically be removed leading to increased line-to-line capacitance and via resistance adversely affecting device performance and functionality. If this modified layer is not removed, there may be potential device reliability issues associated with these

structures. It is, thus, necessary to utilize a resist ash process that induces minimal chemical and physical modification of the OSG sidewall. For this purpose we propose the use of an *in situ* Ar/H₂ plasma on the identical commercial etch platform used for via and trench formation during damascene processing. The Ar/H₂ ash process is typically
5 composed of ~90% H₂ and ~10% Ar. Other typical operating conditions on one specific commercial etch platform are: 1 Torr chamber pressure, 500 sccm H₂ and 50 sccm Ar flow, 600 W 27MHz ("Source") Power and <50W 2MHz ("Bias") Power. The process can yield sufficiently quick strip rates (> 120 nm/min) by operating at elevated pressures (1 Torr) and substrate temperatures with only marginal "bias" (substrate) power applied.

10 Since the plasma in this ash process is ~ 90% hydrogen, hydrogen atoms are the dominant radicals. Two direct routes of OSG material degradation through chemical reactivity include reactivity of exposed Si bonds created during ion impact to the surface and carbon removal from the OSG film. In terms of Si reactivity, the chemisorptive sticking coefficient on Si of hydrogen (<< 0.001) [1] is smaller than that of nitrogen (>
15 0.05) [2] and oxygen (> 0.1) [2]. As such, a hydrogen-based ash process is likely to have reduced Si chemical reactivity with the OSG sidewall. On the issue of carbon removal, the reaction mechanism for the removal of carbon with oxygen and nitrogen based strip processes is via the formation of CO and CN species respectively. Hydrogen based strip process will likely form various CH_x (x=1-3) species which are less likely to be volatilised
20 than CO and CN, thus removing less carbon from the film. Further, on commercial BEOL etch platforms, these ash processes run at pressures > 200 mT, implying that the positive ion mean free path < 0.5 cm [3], [4], i.e. less than the spacing between top and bottom electrodes of most BEOL commercial etch platforms. Ion scattering and, thus, ion impact on the OSG sidewall likely causes physical damage to the OSG material. However, the

dominant ion in these Ar/H₂ plasmas, H⁺ (1 amu), will probably cause less damage to the ILD sidewall than O⁺ (16 amu) or N⁺ (14 amu) because of its much smaller mass.

The aforementioned physical and chemical properties of this *in situ* Ar/H₂ process, thus facilitates minimized dielectric modification during ashing and consequent
5 desirable device characteristics.

- [1] P. Bratu, K. L. Kompa, and U. Höfer "Optical second-harmonic investigations of H₂ and D₂ adsorption on Si(100)2×1: the surface temperature dependence of the sticking coefficient"
10 Chem. Phys. Lett. 251, 1-7 (1996)
[2] D. Graves, Dept. of Chemical Engineering, UC Berkeley, private communication.
[3] N. C. M. Fuller, *Phd Dissertation*, "Controlling the Relative Rates of Adlayer Formation and Removal during Etching in Inductively-Coupled Plasmas" Columbia University (2002)
15 [4] *Principles of Plasma Discharges and Materials Processing*, edited by M. A. Liebermann and A. J. Lichtenberg (Wiley, New York 1994), pg. 47.

20 SUMMARY

It is therefore an object of the present invention to provide a BEOL interconnect structure of, e.g., the dual damascene type, in which the original physical and chemical properties of the dielectric are un-altered during ashing in both the thinwire and fatwire levels.

25 It is a further object of the present invention to provide a BEOL interconnect structure in which both dielectric properties and feature profiles are un-altered during post dual damascene wafer de-fluorination in both thinwire and fatwire levels.

Thus, it is a further object of the present invention to provide a BEOL interconnect structure of improved device functionality, performance, and reliability owing to the reduced demand for removing modified dielectric material.

In keeping with these and other objects of the present invention, there is
5 provided an interconnect structure comprising an organosilicate (OSG) low-k dielectric layer having a set of metallic lines formed therein; such that the surface of the low-k dielectric in contact with the metallic lines is of the original physical and chemical integrity of and matches that of the bulk low-k material facilitating improved device characteristics.

10 BRIEF DESCRIPTION OF THE DRAWINGS

Preferred embodiments of the present disclosure are described below with reference to the drawings, which are described as follows:

FIG. 1 is a cross sectional SEM showing ash-induced modification of JSR
5109 porous OSG dielectric material utilizing a nitrogen-based de-fluorination process.

15 FIG. 2 is a cross sectional SEM showing no ash-induced modification of JSR 5109 porous OSG dielectric material utilizing the proposed Ar/H₂ de-fluorination process.

DESCRIPTION OF THE PREFERRED EMBODIMENT

20 The present invention which is directed to an interconnect structure useful for forming a semiconductor device, the interconnect structure having an unaltered dielectric material physical and chemical properties facilitating improved device performance, functionality, and reliability.

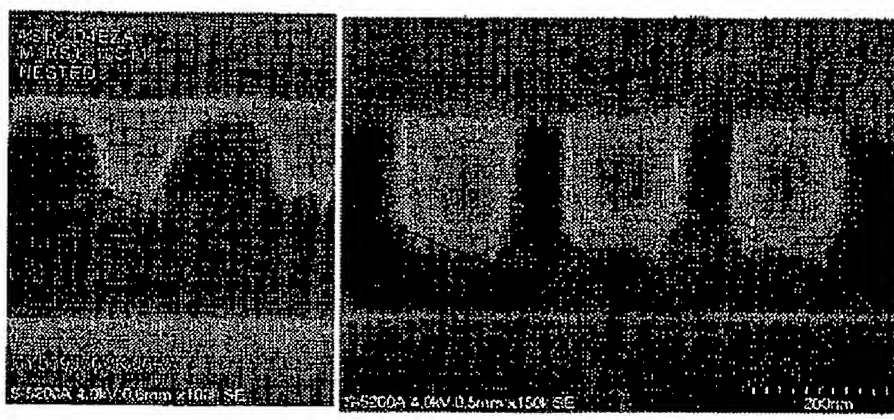
The formation of this structure is made feasible by utilizing an *in situ* Ar/H₂ ash process post via and trench processing in dual damascene scheme that induces minimal chemical and physical modification of the OSG sidewall. The Ar/H₂ ash process is typically composed of ~90% H₂ and ~10% Ar. Other typical operating conditions on one
5 typical commercial etch platform are: 1 Torr chamber pressure, 500 sccm H₂ and 50 sccm Ar flow, 600 W 27MHz ("Source") Power and <50W 2MHz ("Bias") Power. The process yields sufficiently quick strip rates (> 120 nm/min) by operating at elevated pressures (1 Torr) and substrate temperatures with only marginal "bias" (substrate) power applied.

This Ar/H₂ ash process minimizes chemical and physical reactivity with OSG
10 materials owing to its inherent chemical make up: H-radicals are unable to achieve as efficient chemical reactivity with the Si and Carbon within the dielectric film compared with other conventional ash chemistries and the minute ionic mass of H⁺ ions ensures minimal physical damage during ion scattering occurring in the ash process.

As a consequence of such a chemically and physically "friendly" ash process, the
15 original chemical and physical integrity of the OSG material is maintained, eliminating the need for removing a modified OSG layer and, further, facilitating improved device performance, functionality and reliability. Further, the characteristics of this process enables its use for post dual damascene wafer de-flourination (DF).

Figure 1
N₂/H₂ DF

Figure 2
Ar/H₂ DF



ABSTRACT

Novel interconnect structures possessing an organosilicate dielectric material with unaltered physical and chemical material properties post exposure to a specific resist ash chemistry for use in semiconductor devices are provided herein. The
5 novel interconnect structure is capable of delivering improved device performance, functionality and reliability owing to the use of a chemically and physically "friendly" resist ash process. The proposed *in situ* Ar/H₂ process achieves minimal chemical and physical reactivity with the OSG sidewall during ashing owing to its inherent make up; H radicals in the ash chemistry have mild chemical reactivity with the elements in the OSG
10 film and the minute ionic mass of H⁺ ions implies minimal physical bombardment of the OSG film during ashing. The proposed chemistry is also particularly useful for post dual damascene wafer de-fluorination for the aforementioned reasons.

EXHIBIT B

Nicholas C Fuller

To: Robert Wisniewski/Watson/IBM@IBMUS, Robert M
Trepp/Watson/IBM@IBMUS

cc: Kaushik Kumar/Fishkill/IBM@IBMUS, Timothy
Dalton/Fishkill/IBM@IBMUS

This document expires on

From: Nicholas C Fuller/Watson/IBM@ibmus

Subject: Disclosure YOR8-2003-0269

This document contains a file attachment whose file size is: 2.9 MB. Click here to delete all attachments.

Gentlemen

As per our last mtg. to discuss the above disclosure, the patent has been written and is attached, please let me know if you have any questions.

Thanks



Photoresist_Ash_Process.doc

Nicholas C.M. Fuller, Ph.D
Research Staff Member, Plasma Process Development
IBM T.J. Watson Research Center, 01-159
Yorktown Hts, NY 10598
Ph: (845)-892-4174
Alt: (914)-945-2660

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